Solution to problems of Chapter 1

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Relation between the units:

$$\begin{array}{c} 1 \text{ ft=}0.3048\text{m} \\ 1 \text{lb=}0.454 \text{ kg} \\ 1 \text{lb/ft}^2 = 47.89 \text{ N/m}^2 = 47.89 \text{ Pa} \\ 1^o \text{R=}5/9 \text{K} \end{array}$$

1.1 At the nose of a missile in flight, the pressure and temperature are 5.6 atm and 850°R, respectively. Calculate the density and specific volume. (Note: $1 \text{ atm} = 2116 \text{ lb/ft}^2$.)

Solution:

The temperature is $T = 850^{\circ} R = 850 \times 5/9 K = 472.2 K$

The pressure is p=5.6 atm= $5.6 \times 1.01 \times 10^5 \text{ N/m}^2 = 5.656 \times 10^5 \text{N/m}^2$.

The density is $\rho = \frac{p}{RT} = \frac{5.656 \times 10^5}{287 \times 472.2} = 4.1735 \text{ kg/m}^3$. The specific volume is $v = \frac{1}{\rho} = 0.2396 \text{m}^3/\text{kg}$

1.2 In the reservoir of a supersonic wind tunnel, the pressure and temperature of air are 10 atm and 320 K, respectively. Calculate the density, the number density, and the mole-mass ratio. (Note: 1 atm = $1.01 \times 10^5 \text{ N/m}^2$.) Solution:

The density is $\rho = \frac{p}{RT} = \frac{10 \times 1.01 \times 10^5}{287 \times 320} = 10.997 \text{kg/m}^3$. The number density is $n = \frac{p}{kT} = \frac{10 \times 1.01 \times 10^5}{1.38 \times 10^{-23} \times 320} = 2.287 \times 10^{26} / \text{m}^3$. The mole-mass ratio is $\eta = \frac{pv}{\mathscr{R}T} = \frac{p}{\rho \mathscr{R}T} = \frac{10 \times 1.01 \times 10^5}{10.997 \times 8314 \times 320} = 0.0345 \frac{\text{kg} \cdot \text{mol/kg}}{\text{kg}}$.

1.3 For a calorically perfect gas, derive the relation $c_p - c_v = R$. Repeat the derivation for a thermally perfect gas.

Solution:

For calorically perfect gas $h = c_p T$ and $c_v T$.

From the definition of enthalpy, we have $h = e + pv \Rightarrow c_pT = c_vT + RT \Rightarrow c_p - c_v = R$.

For thermally perfect gas dh = de + d(pv), $dh = c_p dT$ and $de = c_v dT$.

Thus we have $c_p dT = c_v dT + d(RT) \Rightarrow c_p = c_v + R \Rightarrow c_p - c_v = R$.

1.4 The pressure and temperature ratios across a given portion of a shock wave in air are $p_2/p_1 = 4.5$ and $T_2/T_1 =$ 1.687, where 1 and 2 denote conditions ahead of and behind the shock wave, respectively. Calculate the change in entropy in units of (a) $(ft \cdot lb)/(slug \cdot {}^{o}R)$ and (b) $J/(kg \cdot K)$.

Solution:

The change in entropy is given by

$$\Delta S = c_p \ln \frac{T_2}{T_1} - R \ln \frac{p_2}{p_1} = \frac{\gamma R}{\gamma - 1} \ln \frac{T_2}{T_1} - R \ln \frac{p_2}{p_1}$$

1.5 Assume that the flow of air through a given duct is isentropic. At one point in the duct, the pressure and temperature are $p_1 = 1800 \text{ lb/ft}^2$ and $T_1 = 500 \text{ }^o\text{R}$, respectively. At a second point, the temperature is 400 ^oR . Calculate the pressure and density at this second point.

Solution:

For isentropic process, we have

 $\frac{p_2}{p_1} = (\frac{\rho_2}{\rho_1})^{\gamma} = (\frac{T_2}{T_1})^{\gamma/(\gamma-1)}$

Thus

 $p_2 = p_1(\frac{T_2}{T_1})^{\gamma/(\gamma-1)} = 824.30lb/ft^2$

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$$\rho_2 = \frac{p_2}{RT_2} = 0.6179kg/m^3$$

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1.6 Consider a room that is 20 ft long, 15 ft wide, and 8 ft high. For standard sea level conditions, calculate the mass of air in the room in slugs. Calculate the weight in pounds. (Note: If you do not know what standard sea level conditions are, consult any aerodynamics text, such as Refs. 1 and 104, for these values. Also, they can be obtained from any standard atmosphere table.)

Solution:

 $V = 20 \times 15 \times 8 = 2400 ft^3$

 $\rho = 1.225 kg/m^3 = 2.4 \times 10^{-3} slug/ft^3$ for standard sea level condition.

Thus $m = \rho V = 5.76 slug = 5.76 \times 14.5935/0.454lb = 185.15lb$.

1.7 In the infinitesimal neighborhood surrounding a point in an inviscid flow, the small change in pressure, dp, that corresponds to a small change in velocity, dV, is given by the differential relation $dp = -\rho V dV$. (This equation is called Euler's Equation; it is derived in Chap. 6.) a. Using this relation, derive a differential relation for the fractional change in density, $d\rho/\rho$, as a function of the fractional change in velocity, dV/V, with the compressibility τ as a coefficient.

b. The velocity at a point in an isentropic flow of air is 10 m/s (a low speed flow), and the density and pressure are 1.23 kg/m^3 and $1.01 \times 10^5 \text{ N/m}^2$, respectively (corresponding to standard sea level conditions). The fractional change in velocity at the point is 0.01. Calculate the fractional change in density.

c. Repeat part (b), except for a local velocity at the point of 1000 m/s (a high-speed flow). Compare this result with that from part (b), and comment on the differences.

Solution:

(a)

$$\tau = \frac{1}{V} \frac{dV}{dp} \tag{1}$$

and

$$V = \frac{1}{\rho} \Rightarrow dV = \frac{1}{\rho^2} d\rho \tag{2}$$

Substitute equation (2) into equation (1), we have

$$\tau = \frac{1}{\rho^2 V} \frac{d\rho}{dp} \tag{3}$$

. Substitute the Euler's equation into equation (3), we have

$$\tau = -\frac{1}{\rho} \frac{d\rho}{\rho V dV} \Rightarrow \frac{d\rho}{\rho} = -\tau \rho V^2 \frac{dV}{V} \tag{4}$$

(b). For isentropic flow, we have

$$\frac{p}{\rho^{\gamma}} = c \Rightarrow p = c\rho^{\gamma} \Rightarrow dp = \gamma c\rho^{\gamma - 1} d\rho = \gamma c \frac{\rho^{\gamma}}{\rho} = \gamma \frac{p}{\rho} d\rho \Rightarrow \frac{dp}{d\rho} = \gamma \frac{p}{\rho}$$
 (5)

$$\tau = \frac{1}{\rho} \frac{dp}{d\rho} = \frac{1}{\gamma p} \tag{6}$$

Substitute equation (6) into equation (4), we have

$$\frac{d\rho}{\rho} = -\frac{1}{\gamma p} \rho V^2 \frac{dV}{V} \tag{7}$$

For V=10m/s, p=1.01×10⁵N/m², $\rho=1.23$ kg/m³, $\gamma=1.4$ and $\frac{dV}{V}=0.01$, we have $\frac{d\rho}{\rho}=-8.99\times10^{-6}$. (c) For V=1000m/s, we have $\frac{d\rho}{\rho}=-8.99\times10^{-2}$ It can be seen that the larger the velocity, the larger the relative change in density.